## Chemistry 125 Seventh Examination April 13, 2007

Name \_\_\_\_\_

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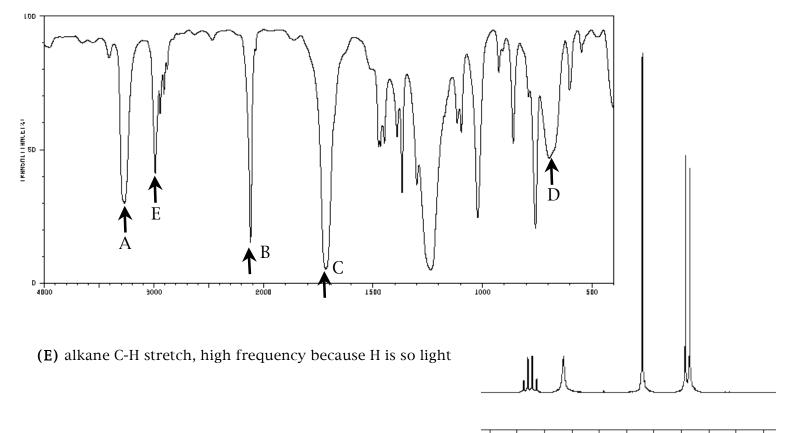
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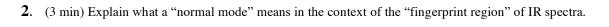
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O OH

1. (12 min) One of spectra below is for ethyl propiolate, H-C=C-C-O-CH<sub>2</sub>-CH<sub>3</sub>, the other for acetoin, CH<sub>3</sub>-C-CH-CH<sub>3</sub>.

\*\*Label\* each spectrum with the compound name \*\*and explain\* with just a few words \*\*every\* significant PMR peak and the four IR peaks labeled \*\*A-D. (E) is given as an example.





- **3.** (9 min) Suggest reagent(s) to achieve each of the following purposes: [Just list reagent(s) NO mechanism required]
- a) converting an internal alkyne into a cis double bond
- b) converting an internal alkyne to a terminal alkyne
- c) converting a terminal alkyne into a ketone
- d) converting a terminal alkyne into an aldehyde
- e) converting a C=C double bond into a C=O double bond
- f) converting a C=O double bond into a C=C double bond
- **4.** (4 min) Explain **why** different kinds of magnetic fields are appropriate for chemical NMR and medical MRI.

- 5. (5 min) Answer A OR B, NOT BOTH
- (A) In studying the mechanism of lanosterol biosynthesis using NMR and <sup>13</sup>C double-labeled isopentenyl pyrophosphate, why is it crucial that most of the isopentenyl pyrophosphate not be labeled at all? That is, *why* was *dilute* double labeling used?

(B) Explain why this molecule on the right is well suited to be the visual pigment.

**6.** (7 min) Show the mechanism for acid-catalyzed hydrolysis of the following molecule to 1,2-ethanediol and acetone. **Use curved arrows**. Several steps are required



7. (10 min) Historically organic chemists speak of "electrophilic" addition of CCl<sub>2</sub> to the C=C group of an alkene and "nucleophilic" addition of CH<sub>3</sub>Li to the C=O group of a ketone. Draw pictures to explain in terms of the *shapes* of *reagent* HOMOs and LUMOs, and *transition state structures*, how these processes are fundamentally similar.